

C16-M/CHOT/RAC-104

6054

BOARD DIPLOMA EXAMINATION, (C-16) OCT/NOV-2018 DME-FIRST YEAR EXAMINATION

ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES

Time: 3 hours | [Total Marks: 80

PART—A

 $3 \times 10 = 30$

Instructions: (1) Answer **all** questions.

- (2) Each question carries three marks.
- (3) Answers should be brief and straight to the point and shall not exceed *five* simple sentences.
- 1. State and explain Aufbau's principle.
- **2.** Draw the shapes of s, p and d orbitals.
- **3.** Define mole. Calculate the number of moles in 200 g of CaCO₃?
- **4.** Define buffer solution. Write any two applications of buffer solutions.
- **5.** Define conductors, electrolytes and nonelectrolytes.

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- **6.** Write the formulas of salts causing hardness of water.
- **7.** Define elastomers. Write any two uses of (a) Bura-s and (b) Neoprene rubber.
- **8.** Write any three characteristics of good fuel.
- **9.** Define primary and secondary air pollutants. Give one example of each.
- **10.** Define renewable and nonrenewable energy sources. Give one example of each.

PART—B

 $10 \times 5 = 50$

Instructions: (1) Answer any **five** questions.

- (2) Each question carries ten marks.
- (3) The answers should be comprehensive and the criterion for valuation is the content but not the length of the answer.
- **11.** (a) State the postulates of Bohr's atomic model.
 - (b) Define covalent bond. Explain the formation of O_2 and N_2 .
- **12.** (a) Define normality. Calculate the equivalent weight of H_2SO_4 , Na_2CO_3 [GMW of H_2SO_4 98 and GMW of Na_2CO_3 106]
 - (b) Explain Bronsted–Lowry theory of acids, bases, neutralization and give two limitations.

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- 13. (a) Explain the process of forth floatation a with neat diagram.
 - (b) Define roasting, calcination and smelting. Give one example each.
- 14. (a) State and explain Faraday's laws of electrolysis.
 - (b) Calculate the amount of copper deposited if 10 ampere current passes through its salt solution for 10 minutes. [Atomic weight of Cu is 63.5 and valency Z=2]
- **15.** (a) What is rusting of iron? Explain the mechanism of rusting of iron with chemical equations.
 - (b) Explain metallic coatings and nonmetallic coatings.
- **16.** (a) Explain softening of hard water by permutit method with a neat diagram.
 - (b) Define reverse osmosis. Write any three advantages of reverse osmosis.
- **17.** (a) Write any five differences between thermoplastics and thermosetting plastics.
 - (b) Write any five advantages of plastics over traditional materials.
- **18.** (a) Define air pollution. Explain any five causes for air pollution.
 - (b) Write a short note on acid rain.

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