MODEL PAPER

Subject Code: R161105/R16

I B. Tech I Semester Regular Examinations Nov. - 2016 **ENGINEERING CHEMISTRY**

(Common to AE,BioTech,ChemE,CE,MinE,MetalE,PE,PCE,AME,ME)

Time: 3 hours

Max. Marks: 70

Question Paper Consists of Part-A and Part-B Answering the question in **Part-A** is **Compulsory**, Four Questions should be answered from Part-B

PART-A

- 1. (a) Discuss the preparation of Thiokol.
 - (b) Define HCV and LCV.
 - (c) Differentiate reversible and irreversible cells.
 - (d) State any four important properties of fullerenes.
 - (e) Write briefly about breakpoint chlorination.
 - (f) What is viscosity index of lubricating oil?
 - (g) Write the anode and cathodic reactions occurring in CH_3OH-O_2 fuel cell.

PART-B

- 2. (a) Discuss (i) emulsion polymerization (ii) p-conducting polymers.
 - (b) Explain compounding of plastics.
- 3. (a) Differentiate octane and cetane number.
 - (b) Calculate the higher and lower calorific value of a fuel that contains 85% carbon, 1.5%sulphur, 0.6% nitrogen, 5.5% hydrogen and 7.4% oxygen. (Latent heat of steam is 587 cal/grams).
 - (c) Explain fixed bed catalytic cracking method for synthesis of petrol.
- 4. (a) Explain the construction and working of dry cell.
 - (b) Explain (i) Pitting corrosion (ii) Impressed current cathodic protection (iii) Electroless plating
- 5. (a) Explain sol-gel method of preparing nano materials.
 - (b) Discuss the types of super conductors.
 - (c) Explain any one method of green synthesis.
- 6. (a) Explain electro-dialysis method for desalination of water.
 - (b) Discuss the troubles caused by boiler scales and how can they be minimized.
 - (c) A sample of hard water gives the following results on analysis: $Ca(HCO_3)_2 16.2$ ppm, Mg(HCO₃)₂ - 14.6 ppm, CaCl₂ - 22.2 ppm, MgCl₂ - 9.5 ppm, CaSO₄ - 13.6 ppm and MgSO₄ – 12 ppm. Calculate the lime and soda required for softening 10,000 litres of this water.

[5+4+5]

- 7. (a) Explain setting and hardening of cement.
 - (b) Write notes on (i) Refractoriness under load (ii) Extreme pressure lubrication.

[7+7]

***** Page 1 of 1

Set No - 1

[8+6]

[7 x 2 = 14]

[4+4+6]

[5+5+4]

[5+9]