

C16-C/CM-104

6019

BOARD DIPLOMA EXAMINATION, (C-16) SEPTEMBER/OCTOBER - 2020 DCE—FIRST YEAR EXAMINATION

ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES (104)

Time: 3 hours [Total Marks: 80

PART—A

 $3 \times 10 = 30$

Instructions: (1) Answer **all** questions.

- (2) Each question carries three marks.
- 1. Write any three differences between orbit and orbital.
- **2.** Define atomic number and mass number. Give one example for each.
- **3.** Define the terms solution, solute and solvent.
- **4.** Define conjugate acid-base pair. Explain with one example.
- **5.** Write any three differences between metallic conduction and electrolytic conduction.
- **6.** Write three essential qualities for drinking water.
- **7.** Define addition polymerization. Give two examples.

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8.	Write the composition of producer gas, water gas and biogas.					
9.	Define COD and BOD.					
10.	Def	Define producers and consumers. Give one example for each.				
		PART—B 10×5=	50			
Instructions: (1) Answer any five questions.						
		(2) Each question carries ten marks.				
11.	(a)	State five postulates of Bohr's atomic theory.	5			
	(b)	Write any five differences between ionic compounds and covalent compounds.	5			
12.	(a)	Define molarity. Calculate the weight of Na $_2$ CO $_3$ present in 500 ml of 0·1 M solution of Na $_2$ CO $_3$. (GMW of Na $_2$ CO $_3$ 106)	5			
	(b)	Define Arrhenius theory of acids and bases, and give three limitations.	5			
13.	(a)	Write any six differences between metals and non-metals.	6			
	(b)	Explain electrolytic refining of metal.	4			
14.	(a)	Explain Faraday's laws of electrolysis.	5			
	(b)	Write any five differences between electrolytic cell and galvanic cell.	5			
15.	(a)	Describe the formation of composition cell, concentration cell and stress cell.	6			
	(b)	Explain the impressed voltage method to control corrosion.	4			
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16.	(a)	Explain ion exchange method to soften hard water.	6
	(b)	Define reverse osmosis. Give any four applications.	4
17.	(a)	Write the preparation and uses of the following with chemical equations: (i) Bakelite (ii) Buna-S-rubber	6
	(b)	Write any four properties of plastics.	4
18.	(a)	Explain any three methods of control of air pollution.	6
	(b)	Write a short note on ozone layer depletion.	4

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