

C16-EE-104/C16-CHPP-104

6037

BOARD DIPLOMA EXAMINATION, (C-16) OCT/NOV-2018 DEEE-FIRST YEAR EXAMINATION

ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES

Time: 3 hours | [Total Marks: 80

PART—A

 $3 \times 10 = 30$

Instructions: (1) Answer **all** questions.

- (2) Each question carries three marks.
- (3) Answers should be brief and straight to the point and shall not exceed *five* simple sentences.
- 1. Write any three differences between orbit and orbital.
- **2.** Write any three characteristics of ionic compounds.
- **3.** Define saturated, unsaturated and supersaturated solutions.
- **4.** Define buffer solution. State their applications.
- **5.** What is electrochemical series? Give its significance.
- **6.** Write any three disadvantages of hard water in industries.
- **7.** Write any three characteristic properties of plastics.

/6037 1 [Contd...

- **8.** Define primary and secondary fuels. Give examples.
- **9.** Write any three threats to biodiversity.
- 10. Write a short note on greenhouse effect.

PART—B

 $10 \times 5 = 50$

Instructions: (1) Answer any **five** questions.

- (2) Each question carries ten marks.
- (3) The answers should be comprehensive and the criterion for valuation is the content but not the length of the answer.
- **11.** (a) Explain:
 - (i) Aufbau's principle
 - (ii) Hund's rule with suitable examples
 - (b) Define unit cell. Draw the structures of unit cell of NaCl and CsCl.
- **12.** (a) Define Normality. Calculate the normality of 250 ml of solution containing 9.8 grams H₂SO₄ (M. Wt of H₂SO₄ 98)
 - (b) Explain Lewis theory of acids and bases.
- **13.** (a) Write any five characteristics of metals.
 - (b) Explain purification of metals by electrolytic refining.
- **14.** (a) Explain the construction and working of galvanic cell with neat diagram.
 - (b) Explain Faraday's Laws of electrolysis.
- **15.** (a) What is rusting of iron? Explain the rusting of iron with chemical equations.
 - (b) Explain impressed voltage method.

/6037 2 [Contd...

- **16.** (a) What are the salts causing temporary and permanent hardness? Give their formulae.
 - (b) Explain softening of hard water by ion-exchange method.
- **17.** (a) Write the preparation and uses of Buna-S and Neoprene rubber.
 - (b) Write any five differences between thermoplastics and thermosetting plastics.
- **18.** (a) Define water pollution. Explain any four causes of water pollution.
 - (b) Write short note on:
 - (i) Acid rains
 - (ii) Ozone layer depletion.

* * *

/6037 3 AA8