

7019

BOARD DIPLOMA EXAMINATION, (C-20) JUNE/JULY—2022

DCE - FIRST YEAR EXAMINATION

ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES

Time: 3 hours [Total Marks: 80

PART—A

 $3 \times 10 = 30$

- **Instructions**: (1) Answer **all** questions.
 - (2) Each question carries **three** marks.
 - (3) Answers should be brief and straight to the point and shall not exceed five simple sentences.
 - 1. What the are fundamental particles in an atom and who invented them?
 - 2. Define molarity. Write the mathematical formula and units for it.
 - 3. Write any three applications of Buffer solution.
 - Define electrolytes and non-electrolytes and also give example for each.
 - 5. Write any three disadvantages of using hard water in industries.
 - 6. State the advantages of plastic over traditional material.
 - Define the fuel. Write the classification of fuels. 7.
 - 8. What is active charcoal? Give some examples.
 - Write the causes of deforestation. 9.
 - 10. Define producers and consumers. Give example for each.

/7019 1 [Contd... **PART—B** 8×5=40

Instructions: (1) Answer **all** questions.

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- (2) Each question carries eight marks.
- (3) Answers should be comprehensive and criterion for valuation is the content but not the length of the answer.
- **11.** (a) Write the differences between ionic and covalent compounds.

(OR)

- (b) State and explain Hund's rule and Pauli's exclusion principle.
- **12.** (a) Define atomic weight, molecular weight, equivalent weight and mole. Give two examples for each.

(OR)

- (b) Explain Lewis acid-base theory with limitations.
- 13. (a) Discuss about Calcination and Roasting with examples.

(OR)

- (b) What is electrolytic cell? Write the differences between electrolytic cell and galvanic cell.
- **14.** (a) What is rusting of iron. Write mechanism of rusting of iron and write any four factors that influence the rusting of iron.

(OR)

- (b) Define soft water. Write the qualities of drinking water.
- **15.** (a) What are elastomers? Write the preparation and uses of Buna S rubber and Neoprene rubber.

(OR)

(b) Define water pollution. Write the causes and controlling methods of water pollution.

/7019 2 [Contd...

PART—C $10 \times 1 = 10$

Instructions: (1) Answer the following question.

- (2) The question carries ten marks.
- (3) Answers should be comprehensive and criterion for valuation is the content but not the length of the answer.
- **16.** What is electrolysis? Analise the products formed at cathode and anode with electrode reactions during electrolysis of aqueous NaCl with a neat diagram.

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