# 7051

## **BOARD DIPLOMA EXAMINATION, (C-20)**

### MAY-2023

## **DME - FIRST YEAR EXAMINATION**

### ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES

Time: 3 Hours [ Total Marks: 80

## PART—A

 $3 \times 10 = 30$ 

**Instructions:** (1) Answer **all** questions.

- (2) Each question carries three marks.
- (3) Answers should be brief and straight to the point and shall not exceed five simple sentences.
- 1. Draw the shapes of 'P' orbital.
- **2.** Calculate the number of moles present in 32 gr of  $O_2$  gas.
- **3.** What is conjugate acid base pair? Give an example.
- **4.** Define electrolyte and non-electrolyte. Give examples for each.
- **5.** State any three salts responsible for permanent hardness of water.
- **6.** Define thermosoft plastic. Give one example.
- **7.** What is activated charcoal? Write any two applications of it.
- **8.** State any three characters of a good fuel.
- **9.** State any three causes of water pollution.
- **10.** Define pollutant, contaminant and sink.

**PART—B** 8×5=40

**Instructions:** (1) Answer **all** questions.

- (2) Each question carries eight marks.
- (3) Answers should be comprehensive and criterion for valuation is the content but not the length of the answer.
- **11.** Explain the significance of four quantum numbers.

(OR)

Define Ionic bond and Covalent bond. Write any four differences between Ionic and Covalent compounds.

**12.** Define normality. Calculate the normality of solution prepared by dissolving 5.3 gr of Na<sub>2</sub>CO<sub>3</sub> in 500 ml solution.

(OR)

Explain Arrhenius theory of acids and bases with examples. Mention any two limitations.

- **13.** Define the following and give one example for each:
  - (a) Mineral
  - (b) Ore
  - (c) Flux
  - (d) Alloy

(OR)

Explain the structure and working of galvanic cell with a neat diagram.

**14.** What is cathodic protection? Explain sacrificial anode method with a neat diagram.

(OR)

Explain permutit process of softening of hard water with a diagram.

**15.** What is meant by elastomer? Explain the preparation of BUNA-S with equation and mention two applications.

(OR)

Write short notes on (a) deforestation and (b) greenhouse effect.

**Instructions:** (1) Answer the following question.

- (2) The question carries **ten** marks.
- (3) Answer should be comprehensive and the criterion for valuation is the content but not the length of the answer.
- **16.** Explain vulcanization of natural rubber with chemical equation. Write any three properties of vulcanized rubber.

