

Code No: 121AE

JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD**B.Tech I Year Examinations, August/September - 2016****ENGINEERING CHEMISTRY****(Common to all Branches)****Time: 3 hours****Max. Marks: 75****Note:** This question paper contains two parts A and B.

Part A is compulsory which carries 25 marks. Answer all questions in Part A. Part B consists of 5 Units. Answer any one full question from each unit. Each question carries 10 marks and may have a, b, c as sub questions.

PART- A**(25 Marks)**

- 1.a) What the difference between primary cell and secondary cell? [2]
- b) What is water line corrosion? Explain with an example. [3]
- c) What are plastics, elastomers, fibers? Give examples for each. [2]
- d) Write about chemical vapors deposition in the preparation of nano materials. [3]
- e) Give the units in which hardness of water is expressed. Give their inter relation. [2]
- f) Write about ozonisation and its significance. [3]
- g) Define and explain octane number. [2]
- h) Give the significance of flue gas analysis. [3]
- i) What are differences between melting point, triple point and eutectic point in phase diagram? [2]
- j) Give an account on applications of adsorption. [3]

PART-B**(50 Marks)**

- 2.a) Give the expression for Nernst equation. Mention its applications.
- b) What are various types of dry corrosion? Explain with suitable examples.
- c) Write an account on cementation and galvanizing. [3+4+3]

OR

- 3.a) Give the construction and working of glass electrode.
 - b) What are paints .Give the constituents and functions of any four constituents.
 - c) Explain the construction and working of H₂O₂ fuel cell. What are its advantages? [3+3+4]
- 4.a) Give the characteristics of fibers. What are the applications of FRPS?
 - b) Give the mechanism of thick film lubrication.
 - c) Mention the preparation, properties and engineering applications of PVC and butyl rubber. [3+3+4]

OR

- 5.a) Give the mechanism of conduction in poly acetylene.
- b) What are nanomaterials? Give their applications.
- c) How do refractories are classified. Give examples. [3+4+3]

- 6.a) How do you determine hardness of water by EDTA complexometric method.
b) What are steps involved in the treatment of domestic water. [5+5]

OR

- 7.a) Discuss about internal treatment methods of boiler feed water .
b) How disinfection of water is carried out by ozonisation and chlorination. [5+5]

- 8.a) How do you analyze coal by ultimate analysis. Give its significance.
b) Calculate the mass of air required for the complete combustion of 5.0 kg. of coal containing 80 % of carbon, 15% hydrogen and the rest oxygen. [4+6]

OR

- 9.a) A sample of coal was found to contain the following percentage composition of C = 75%, H = 5.2% , O = 12%, N = 3.2% , ash = 4.5% calculate the minimum air required for complete combustion of 1kg of coal.

- b) Explain Fischer – Tropsch’s process with a neat diagram.
c) What are the advantages of gaseous fuels over liquid fuels? [4+4+2]

- 10.a) What are colloids? How are they classified?
b) Explain the salient features of physical adsorption and chemical adsorption. [3+4+3]
c) Explain about hardening and normalization of steel.

OR

- 11.a) Discuss the application of phase rule to Ag – Pb system.
b) Differentiate physisorption and chemisorptions. [5+5]

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